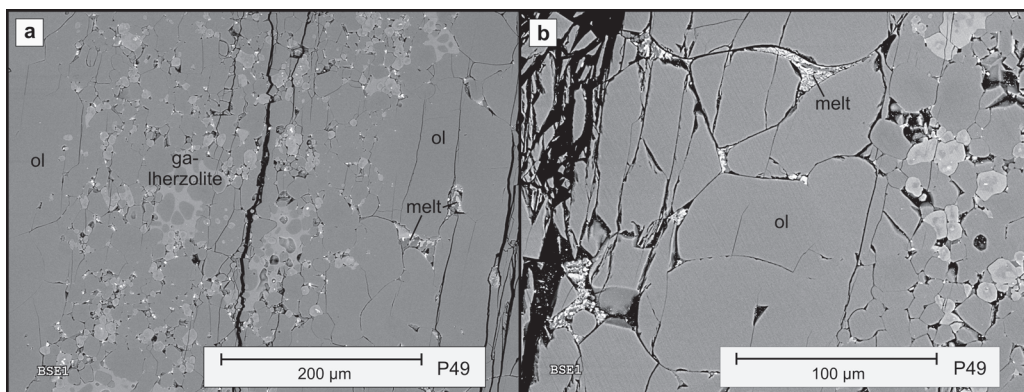


Supplementary Figure S1 | Scanning electron microscope images of experimental charges with vapour and vapour phase quench. Experimental runs are in the order of added water concentration from the lowest wt.% of added H₂O to highest wt.% of added H₂O. parg – pargasite, ga – garnet, cpx – clinopyroxene, opx – orthopyroxene, ol – olivine.

- A) Experiment D968 (P30) HZ1 lherzolite composition at 2.5 GPa, 1000°C, dry (i.e. no H₂O added). Back-scattered electron image of polished surfaces of pargasite lherzolite. The order of decreasing 'brightness' in the images is clinopyroxene > garnet (poikilitic) > pargasite > olivine > orthopyroxene, similar as in B) and C). Pargasite grains are right of the centre. Although nominally 'dry', the presence of pargasite indicates that trace water/hydrogen gained access. Also note the lower intergranular void space in A) and B) relative to D) (1.45 wt.% H₂O).
- B) Experiment C2901 (O57) HZ2 lherzolite composition at 2.5 GPa, 1000°C, 0.073 wt.% H₂O. Back-scattered electron image of polished surfaces of pargasite lherzolite. Pargasite grains are in the upper right of the image.
- C) Experiment D1036 (P77(2)) HZ1 lherzolite composition at 1.5 GPa, 950°C, ~1 wt.% H₂O. Secondary electron image of lherzolite with excess water-rich vapour showing pargasite intergrowth with olivine and pyroxenes. Broken surface illustrating crystal faces against voids (vapour). Pargasite in the upper centre.
- D) Experiment C2886 (O52) HZ2 composition at 2.5 GPa, 1025°C, 1.45 wt.% H₂O. Back-scattered electron image of polished surfaces of pargasite lherzolite. Pargasite grains are in the upper right of the image. Pargasite increases in abundance A, B, D as shown by decreasing K₂O in pargasite.
- E) Experiment C2876 (O49) HZ2 lherzolite composition at 2.5 GPa, 1025°C, 1.45 wt.% H₂O. Back-scattered electron image of edge of pargasite-bearing garnet lherzolite layer against the layer of carbon spheres. Silicate minerals between spheres are not quenched from silicate melt or vapour phase but are pargasite (the most common phase), clinopyroxene, orthopyroxene, garnet and olivine with compositions identical with those of the lherzolite layer, except for vanishingly small Cr₂O₃ contents in all phases. The minerals are interpreted as void-filling growth of equilibrium phases by element transport through the vapour phase.
- F) Experiment C2919 (O89) HZ2 lherzolite composition at 2.5 GPa, 1000°C, 7.25 wt.% H₂O. Back-scattered electron image of fragment of olivine disc within garnet lherzolite. The olivine single crystal shows an open fracture along which olivine is replaced by clinopyroxene giving a discontinuous clinopyroxene veinlet. Olivine close to the fracture or disc edge is ~Mg_{93.5} whereas olivine at >10 microns from the fracture has Mg₉₁. The insert is a secondary electron image of a train of fluid inclusions within the same olivine disc.

- G) Experiment C2836 (O18) HZ2 lherzolite composition at 2.4 GPa, 960°C, 14.5 wt.% H₂O. Euhedral olivine crystals with fragmented thin films of silica-rich glass (secondary electron image).
- H) Experiment C2841 (O22) HZ2 lherzolite composition at 2.8 GPa, 840°C, 14.5 wt.% H₂O. Quench 'rosette' of carbonate and glass fragments adhering to euhedral pyroxene and olivine crystals (secondary electron image).



Supplementary Figure S2 | Scanning electron microscope images of experimental charges with quench from water-rich silicate melt of experiment UHPPC114 (P49) HZ1 lherzolite composition at 6 GPa, 1400°C, 1.45 wt.% H₂O. Abbreviations as in Supplementary Figure S1.

- A) Back-scattered electron image of lherzolite layer between two olivine layers. Recrystallization and grain growth of olivine is accompanied by migration of hydrous silicate melt into the 'olivine melt trap'.
- B) Back-scattered electron image of olivine layer with lherzolite layer. Image shows interserts of quench clinopyroxene, oxide and carbonate, and quench rims on olivine.

Supplementary Table S3 | Experimental conditions and results on 'HZ1' composition. HZ1 and MOR Pyrolite are estimates of the upper mantle composition in the region acting as the source for MORB i.e. the asthenosphere.

Run No.	Mount No.	T (°C)	Time (days)	wt.% H ₂ O	Capsule	Top L	Bottom L	Phase assemblage (with OL+OPX+CPX)			
								SP/GA	PAR	Melt	Vap
<i><u>HZ1 Peridotite at 1.5 GPa</u></i>											
D1036	P77(2)	950	4	~1	Au	-	-	SP	PAR	-	Vap
D1041	P82	1025	3	~0.05	Au	OL(SC)	OL(SC)	SP	PAR	-	-
C3230	P83	1025	3	~0.3	Au	OL(SC)	OL(SC)	SP	PAR	-	-
<i><u>HZ1 Peridotite at 2.5 GPa</u></i>											
D968	P-30	1000	7	DRY	Au	OL(SC)	Lo-Al Opx	GA	PAR	-	-
D1039	P80	1000	3	~0.3	Au	OL(SC)	OL(SC)	GA	PAR	-	-
D897	O-85	1000	2.7	1.45	Ag	OL(SC)	OL(SC)	GA	PAR	-	Vap
D937	P-4	1000	7	1.45	Au	Hi-Al Opx	Hi-Al Cpx	GA	-	-	Vap
D944	P-7	1000	7	1.45	Au	Lo-Al Opx	Lo-Al Cpx	GA	-	-	Vap
C3225	P81	1025	3	~0.3	Au	OL(SC)	OL(SC)	GA	PAR	Melt ^a	
C2936	O-79	1025	3	1.45	Ag	OL(SC)	OL(SC)	GA	-	Melt ^a	Vap
D1038	P78	1050	3	~0.05	Au	OL(SC)	OL(SC)	GA	PAR	-	-
C3222	P79	1050	3	~0.3	Au	OL(SC)	OL(SC)	GA	-	Melt ^a	-
C2930	O-77	1050	3	1.45	Ag	OL(SC)	OL(SC)	GA	-	Melt ^a	-
<i><u>HZ1 Peridotite at 3.0 GPa</u></i>											
D1035	P77(1)	950	4	~0.5	Au	-	-	GA	PAR	-	Vap
<i><u>HZ1 Peridotite at 4.0 GPa</u></i>											
C2987	O-92	1100	4	0.145	Au	OL(SC)	OL(SC)	GA	-	-	Vap
C2942	O-80	1100	3	1.45	Au	OL(SC)	OL(SC)	GA	-	-	Vap
C2935	O-78	1150	3	1.45	Au	OL(SC)	OL(SC)	GA	-	-	Vap

C3005	P-3	1150	7	1.45	Au	Hi-Al Opx	Hi-Al Cpx	GA	-	-	Va _l
C3010	P-6	1150	7	1.45	Au	Lo-Al Opx	Lo-Al Cpx	GA	-	-	Va _l
C2950	O-81	1200	3	1.45	Au	OL(SC)	OL(SC)	GA	-	-	Va _l
C2899	O-98/99	1225	1	1.45	AuPd	OL(SC)	OL(SC)	GA	-	Melt ^a	-

HZ1 Peridotite at 6.0 GPa

UHPPC117	P53	1300	12h	1.45	AuPd	OL(SC)	OL(SC)	GA	-	-	?Va
UHPPC116	P51	1350	12h	0.145	AuPd	OL(SC)	OL(SC)	GA	-	-	?Va
UHPPC129	P84	1350	12h	~1	AuPd	1/3 OL		GA	-	-	?Va
UHPPC95	P-21	1350	3	1.45	AuPd	OL(SC)	Lo-Al Opx	GA	-	-	?Va
UHPPC131	P86	1400	12h	~1	AuPd	1/3 OL		GA	-	Melt ^a	-
UHPPC113	P47	1400	12h	1.45	AuPd	OL(SC)	Lo-Al Opx	GA	-	Melt ^a	-
UHPPC114	P49	1400	12h	1.45	AuPd	OL(SC)	OL(SC)	GA	-	Melt ^a	-
UHPPC130	P85	1450	4.5h	~1	AuPd	1/3 OL		GA	-	Melt ^a	-

OL, olivine; OPX, orthopyroxene; CPX, clinopyroxene; SP, Spinel; GA, garnet; PAR, pargasite; Vap, vapour; L, Layer; C-Sph, Carbon-spheres; Lo-Al OPX, Low alumina orthopyroxene; Hi-Al OPX, High alumina orthopyroxene; OL(SC), San Carlos olivine; AuPd, AuPd double capsule; ^a, Quench materials of cpx or amp, oxide and carbonate - hydrous silicate melt quench; ^b, Quench materials of dolomite, minor oxide and minor Na-Si-rich silicate – Fluid phase or carbonatitic melt quench.

Supplementary Table S4 | Experimental conditions and results on 'HZ2' composition.

Run#	Mount#	T (°C)	Time (days)	wt.% H ₂ O	Capsule	Top L	Bottom L	Phase assemblage (with OL+OPX+CPX+GA)		
								PAR	Melt	Vap
<i><u>HZ2 Peridotite at 2.0 GPa</u></i>										
C2866	O-35,38	850	3	14.5	Ag	C-Sph	C-Sph	PAR	-	Vap
C2848	O-26	940	7	1.45	Au	-	-	PAR	-	Vap
<i><u>HZ2 Peridotite at 2.4 GPa</u></i>										
C2842	O-24	960	6	1.45	Au	-	-	PAR	-	Vap
C2836	O-17,18	960	4	14.5	Au	-	-	-	-	Vap
C2835	O-16 ^a	1100	3	DRY	Au	-	-	-	-	-
<i><u>HZ2 Peridotite at 2.5 GPa</u></i>										
C2911	O-65	1000	7	DRY	G-Ag ^b	-	-	PAR	-	-
C2901	O-57	1000	1	0.073	Ag	-	-	PAR	-	-
C2900	O-56	1000	1	0.145	Ag	-	-	PAR	-	?Vap
D879	O-70	1000	1	0.725	Ag	C-Sph	C-Sph	PAR	-	Vap
C2867	O-36	1000	3	1.45	Ag	C-Sph	C-Sph	PAR	-	Vap
C2888	O-60	1000	3	2.9	Ag		OL-Disc	PAR	-	Vap
C2919	O-89	1000	1	7.25	Ag		OL-Disc	-	-	Vap
C2877	O-50	1000	3	14.5	Ag	C-Sph	C-Sph	-	-	Vap
C2876	O-49	1025	3	1.45	Ag	C-Sph	C-Sph	PAR	-	Vap
C2886	O-52	1025	3	1.45	Ag	OL(SC)	OL(SC)	PAR	-	Vap
C2909	O-62	1050	3	1.45	Ag	C-Sph	C-Sph	PAR	-	Vap
C2887	O-53	1050	3	1.45	Ag	OL(SC)	OL(SC)	-	Melt	-
D881	O-72	1075	1	1.45	Ag	C-Sph	C-Sph	PAR	-	Vap
<i><u>HZ2 Peridotite at 2.8 GPa</u></i>										
C2841	O-22,23	840	7	14.5	Au	-	-	-	-	Vap

C2849 O-27 950 7 1.45 Au - - PAR - Vap

HZ2 Peridotite at 3.0 GPa

D880 O-71 1000 3 0.145 Ag - - PAR - ?Vap

C2924 O-73 1000 3 1.45 Ag C-Sph C-Sph PAR - Vap

HZ2 Peridotite at 4.0 GPa

C2889 O-61 1100 3 2.9 Ag OL Disc - - Vap

C2928 O-76 1150 1 1.45 Ag OL(SC) OL(SC) - - Vap

^a, Run 2836 (O16) contains Spinel rather than Garnet; ^b, Graphite in Ag; OL, olivine; OPX, orthopyroxene; CPX, clinopyroxene; GA, garnet; PAR, pargasite; Vap, Vapour; L, Layer; C-Sph, Carbon-spheres; OL(SC), San Carlos olivine.